

Towards two-dimensional BNC architectures on metal surfaces from self-assembled monolayers

**Carolina M. Ibarra-Barreno¹, Tashfeen Zehra¹, Hamoon Hemmatpour¹, Oreste De Luca¹,
Martina Crosta², Davide Bonifazi², Petra Rudolf¹**

¹*Zernike Institute for Advanced Materials, University of Groningen, 9747 AG Groningen, The Netherlands*

²*Institute of Organic Chemistry, Faculty of Chemistry, University of Vienna, 1090 Wien, Austria*

e-mail: c.m.ibarra.barreno@rug.nl

Graphene is a two-dimensional material with sp²-hybridized carbons in a honeycomb arrangement that shows extraordinary properties such as high tensile strength, extraordinary mechanical stiffness, high electron mobility and transparency [1]. However, graphene has no bandgap, which is a limiting factor for using it in electronic devices [2]. Opening of a band gap can be achieved by heteroatom doping. Methods based on molecular self-assembly can lead towards an efficient tailored synthesis of doped graphene for specific applications, depending on the molecular precursor. Here we aim to synthesize doped graphene using borazine derivatives (introducing a boron-nitrogen ring in the place of an aromatic carbon cyclic) [3]. The molecular precursor containing isolated boron-nitrogen rings is deposited on a Cu substrate by self-assembly, then the self-assembled monolayer is polymerized by using UV light to prevent sublimation when thermally converting them into a (doped) graphene layer. Here we show the first results by X-ray photoelectron spectroscopy (XPS) and describe the difficulties encountered in realizing a specific doping depending on the precursor and in developing 2D boron-nitrogen-carbon materials with reproducible and fine-tuned properties for optoelectronic devices.

References:

- [1] Benhamou, M. (2013). *Advances in Carbon Nanomaterials: Science and Applications*, by Nikos Tagmatarchis.
- [2] Novoselov, K. S. (2012). A roadmap for graphene. *Nature*, 192-200.
- [3] Bonifazi, D. F.-G. (2015). Boron–nitrogen doped carbon scaffolding: organic chemistry, self-assembly and materials applications of borazine and its derivatives. *Chemical Communications*, 15222-15236.